

Review of: "A Law for Irreversible Thermodynamics? Synergy Increases Free Energy by Decreasing Entropy"

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Important roles of synergy in natural and social sciences are emphasized, and are explained by the combination of free energy G and entropy S . This is an interesting approach.

It is found that analyses of such combinations were previously described in Jaffe K and Gerardo Febres 2016.

Unfortunately, the relationship of this paper with the previous work is not clear to me. Table 2 is reproduced from the previous work. Table 3 seems to describe content of Table 2 from a different standpoint of view. Probably, Tables 1 and 4 are something added in this paper from the previous work. These discussions, however, are not easy to understand, and need to be modified in a way that they can be understood just reading this paper. Tables 1 and 4 may be combined.