

# Review of: "High-pressure thermal conductivity and compressional velocity of NaCl in B1 and B2 phase"

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The present work brings light on the pressure-induced structural phase transition from phase B1 to B2 in sodium chloride (NaCl) at room-temperature. The author presents very interesting experimental results, which are discussed in terms of the connection between the thermal conductivity and the effective Grüneisen parameter  $\gamma$  via the so-called Leibfried-Schlömann (LS) equation. It is known that the above-mentioned structural phase transition is accompanied by an entropy change of  $(1.5 \pm 0.3) \text{ cal} \cdot \text{deg}^{-1}$  under a pressure of 300 kbar and a volume change of  $(-1.00 \pm 0.05) \text{ cm}^3 \cdot \text{mole}^{-1}$ , cf. Ref. [1]. It has been discussed in the literature that  $\gamma$  and the so-called Grüneisen ratio (GR) are powerful tools to explore both phase transitions and critical phenomena, following discussions presented in Refs [2-8] and references cited therein. This is because  $\gamma$ , as well as the GR, quantifies the entropy change associated with the phase transition of interest upon varying both temperature and the control parameter. In the present case, the author considers  $\gamma$  as a constant at phases B1 and B2 in the analysis of the results. It is of interest to analyze the behavior of  $\gamma$  and GR close to the B1-to-B2 structural phase transition. It is unclear whether such a structural transition is mean-field-type or not. It is well-known that the Ginzburg criterion dictates the region in which fluctuations starts to play a key role [9]. Hence, in the present case, it would be of interest to explore the character of the phase transition, as well as to establish its corresponding Ginzburg criterion. The volume dependence of  $\gamma$  was discussed into details in Ref. [1], namely:

$$(\partial \ln \gamma / \partial \ln V)_T = 1 - \alpha^{-1} (\partial \ln B_T / \partial T)_V - (\partial \ln C_V / \partial \ln V)_T,$$

where  $B_T, C_V$  refer, respectively, to the bulk modulus and the specific heat at constant volume. It turns out that the entropy of the system is changed under 300 kbar as a consequence of the volume change associated with the structural phase transition, cf. Fig. 2 of Ref. [1]. Hence, it is natural to expect that  $\gamma$  varies in the vicinity or at the transition. In other words,  $\gamma$  does not vary too much far from the phase transition, cf. literature results [6]. However, at or close to the B1-to-B2 structural phase transition under 300 kbar, it is expected that  $\gamma$  is no longer constant.

Thus, it would be desirable to explore the behavior of  $\gamma$  closer to the B1-to-B2 structural phase transition in

sodium chloride, which is depicted as the highlighted region in pink color in Figs. 1, 3, 4, and 5 of the present work.

Congratulations to the author for the nice work!

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